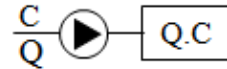


Transfer/Transport in bioreactors

1. Transport: FAST in ONE phase

Feeding, liquid pump, Aeration compressor
Mixing of stirred vessels or bubble columns



2. Transfer: SLOW between TWO phases

O₂ and CO₂ gas-liquid transfer in bioreactor
heat transfer in bioreactor

$$K_L A (C^* - C)$$

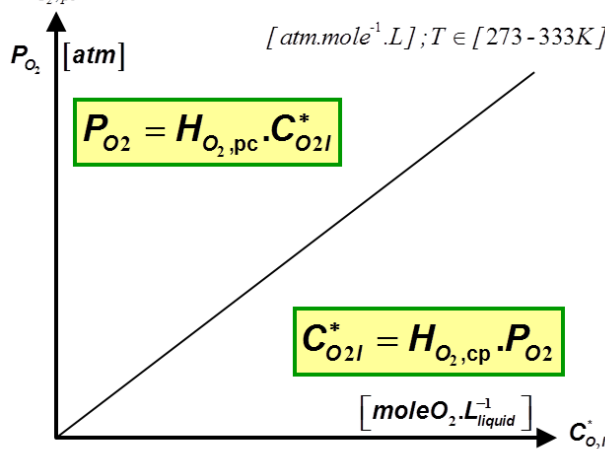
3. Transfer (Fick Diffusion) very SLOW in ONE phase
in biofilms

immobilized enzymes/organisms
in stagnant films at surfaces

$$-AD \frac{dC}{dX}$$

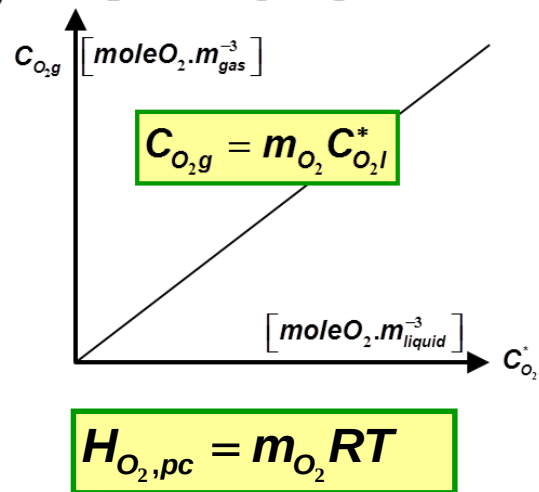
Henry's law H_{O₂}

$$H_{O_2,pc} = \exp(-8391.24/T - 23.24323 \ln(T) + 167.2367)$$



"cp" vs. "pc"
c: O₂ Concentration in liquid
p: O₂ partial Pressure in gas

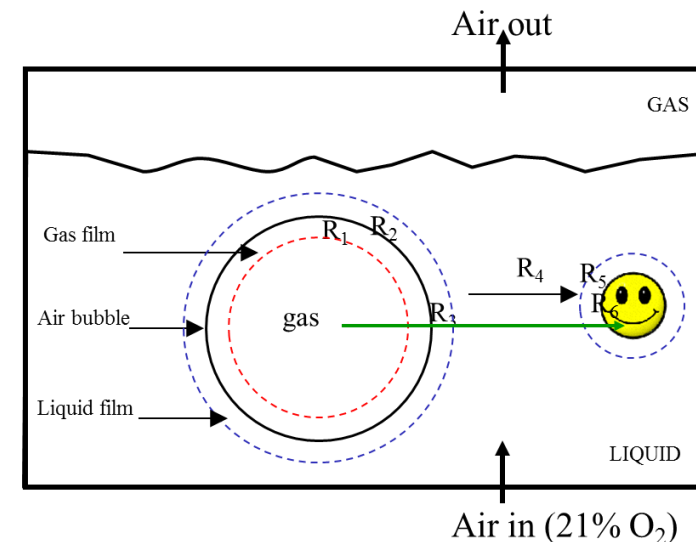
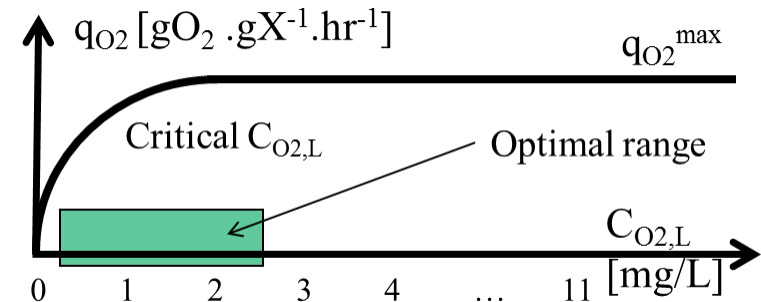
O₂ Gas-liquid partition



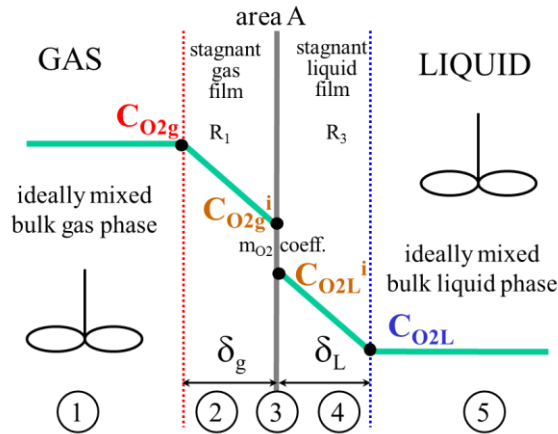
In bioreactor, at steady state

**Biomass
respiration**
 $r_{O_2} = q_{O_2} \cdot X$
[gO₂ . m⁻³ . hr⁻¹]

**Oxygen
transfer rate**
 $OTR = k_L a (*C_{O_2} - C_{O_2,l})$
[gO₂ . m⁻³ . hr⁻¹]



Transfer/Transport in bioreactors



$$\text{rate} = \overbrace{Ak_g (C_{O_2g} - C_{O_2g}^i)}^{\text{gas}} = \overbrace{Ak_l (C_{O_2L}^i - C_{O_2L})}^{\text{liquid}} = \overbrace{K_L A (C_{O_2L}^* - C_{O_2L})}^{\text{overall}} \text{ with } C_{O_2L}^* = m.C_{O_2g}$$

$$\text{rate}_{\text{gas}} = -\frac{AD_g}{\delta_g} (C_{O_2g}^i - C_{O_2g}) = Ak_g (C_{O_2g} - C_{O_2g}^i) \text{ where } k_g \stackrel{\text{def}}{=} \frac{D_g}{\delta_g}$$

$$\text{rate}_{\text{liq.}} = -\frac{AD_L}{\delta_L} (C_{O_2L} - C_{O_2L}^i) = Ak_l (C_{O_2L}^i - C_{O_2L}) \text{ where } k_l \stackrel{\text{def}}{=} \frac{D_L}{\delta_L}$$

$$C_{O_2g}^i = m.C_{O_2L}^i ; mO_2 = 32 \text{ at } 298K$$

$$\frac{1}{K_L} = \frac{1}{k_l} + \frac{1}{m.k_g}$$

$$K_L \approx k_l \stackrel{\text{def}}{=} \frac{D_L}{\delta_L} = \text{in order of } 10^{-4} [\text{m.s}^{-1}]$$

$$OTR = K_L \cdot a (C_{O_2L}^* - C_{O_2L})$$

$$K_L \approx 3 - 4 \times 10^{-4} [\text{m.s}^{-1}] \text{ then if } a \uparrow \Rightarrow K_L a \uparrow$$

$$a = 6 \frac{\varepsilon}{d_b}$$

