Radiation Biology, Protection and Applications (PHYS-450)

Quizzz

Week 10

SOLUTIONS

Problem 1:

Radiation energy spectra can be categorized into two main groups: those that consist of one or more discrete energies (line spectra) and those that consist of a broad distribution of energies (continuous spectra). For each of the radiation sources listed below, indicate whether « line » or « continuous » is a better description:

a) Alpha particles LINE

b) Beta particles CONTINUOUS

c) Gamma rays LINE

d) Characteristic X-rays LINE

e) Conversion electrons LINE

f) Auger electrons LINE

g) Fission fragments CONTINUOUS

h) Bremsstrahlung CONTINUOUS

i) Annihilation radiation LINE

Problem 2:

Which has the higher energy: a conversion electron from the L shell or from the M shell, if both arise from the same nuclear excitation energy?

$$E_{e-}=E_{ex}-E_{binding}$$

 $E_{\text{binding}}(L) > E_{\text{binding}}(M)$

$$E_{e-}(L) < E_{e-}(M)$$

Thus, a conversion electron from the M shell will have higher energy than the electron from the L shell, if both arise from the same nuclear excitation energy.

Problem 3:

Determine ${}_{Z}^{A}X$ in the following nuclear reactions:

a)
$${}_{1}^{2}H + {}_{Z}^{A}X \rightarrow {}_{2}^{4}He + {}_{2}^{4}He$$

b)
$${}_{7}^{14}N + {}_{Z}^{4}X \rightarrow {}_{8}^{17}O + {}_{1}^{1}H$$

$$\circ$$
 ⁴₂He

c)
$${}_Z^A X \rightarrow {}_{27}^{60} Co + \gamma$$

d)
$${}_{2}^{A}X + {}_{2}^{4}He \rightarrow {}_{6}^{12}C^{*} + {}_{0}^{1}n$$