

Radiation Biology, Protection and Applications
(PHYS-450)

EXERCISES

Week 14

Problem 1:

Ingestion of ^{137}Cs eating reindeer meat

In reindeer meat, a concentration of 500 Bq/kg of ^{137}Cs has been measured. What committed effective dose does a person receive on eating 250 grams of this meat?

Problem 2:

Inhalation of ^{131}I

The measured concentration of ^{131}I in a laboratory is 55 Bq/m³. What committed effective dose a person receives during 15 minutes light activity in this laboratory?

Hint: During light work, a reference person inhales 20 liters (0.02 m³) of air per minute. This corresponds to 60 mins. \cdot 0.02 m³/min. = 1.2 m³ per hour. The volume of air inhaled in 15 mins. is then $V = 1.2 \text{ m}^3/\text{h} \cdot 0.25 \text{ h} = 0.3 \text{ m}^3$.

Problem 3:

Inhalation of ^7Be due to BeO from atmosphere

Due to cosmic ray interactions with nitrogen (^{14}N) in the upper atmosphere, each cubic meter of air on the Earth has a concentration of 1 mBq/m³ of radionuclide ^7Be in the form of BeO (beryllium oxide). What is the annual committed effective dose a person receives through this source?

Hint: The inhalation volume of air daily $V_d = 23 \text{ m}^3/\text{day}$ or $V_y = 8400 \text{ m}^3/\text{year}$.