

Radiation Biology, Protection and Applications
(PHYS-450)

Quiz No. 4

Week 13

Problem 1:

The density of an NaCl crystal is 2.17 g/cm^3 . Compute the atom densities of Na and Cl. (The atomic weight of the Na and Cl are 22.990 and 35.453, respectively).

Problem 2:

What is the intensity of gamma rays ($\text{cm}^{-2}\cdot\text{s}^{-1}$) from a 10 curie source of ^{22}Na surrounded by 5 cm of lead? (The build-up factor is 2.1; the ^{22}Na gamma energy is 1.28 MeV and the corresponding total linear absorption coefficient is 0.66 cm^{-1}).

N.B. : $1 \text{ Ci} = 3.7 \times 10^{10} \text{ Bq}$

Problem 3:

Calculate the mass attenuation coefficient of UO_2 for 1MeV γ -rays. (*The density of UO_2 is about 10 g/cm^3 . For 1 MeV γ -rays μ/ρ is $0.0757 \text{ cm}^2/\text{g}$ and $0.0636 \text{ cm}^2/\text{g}$ for uranium and oxygen, respectively).*)

N.B. : Consider ^{238}U in UO_2