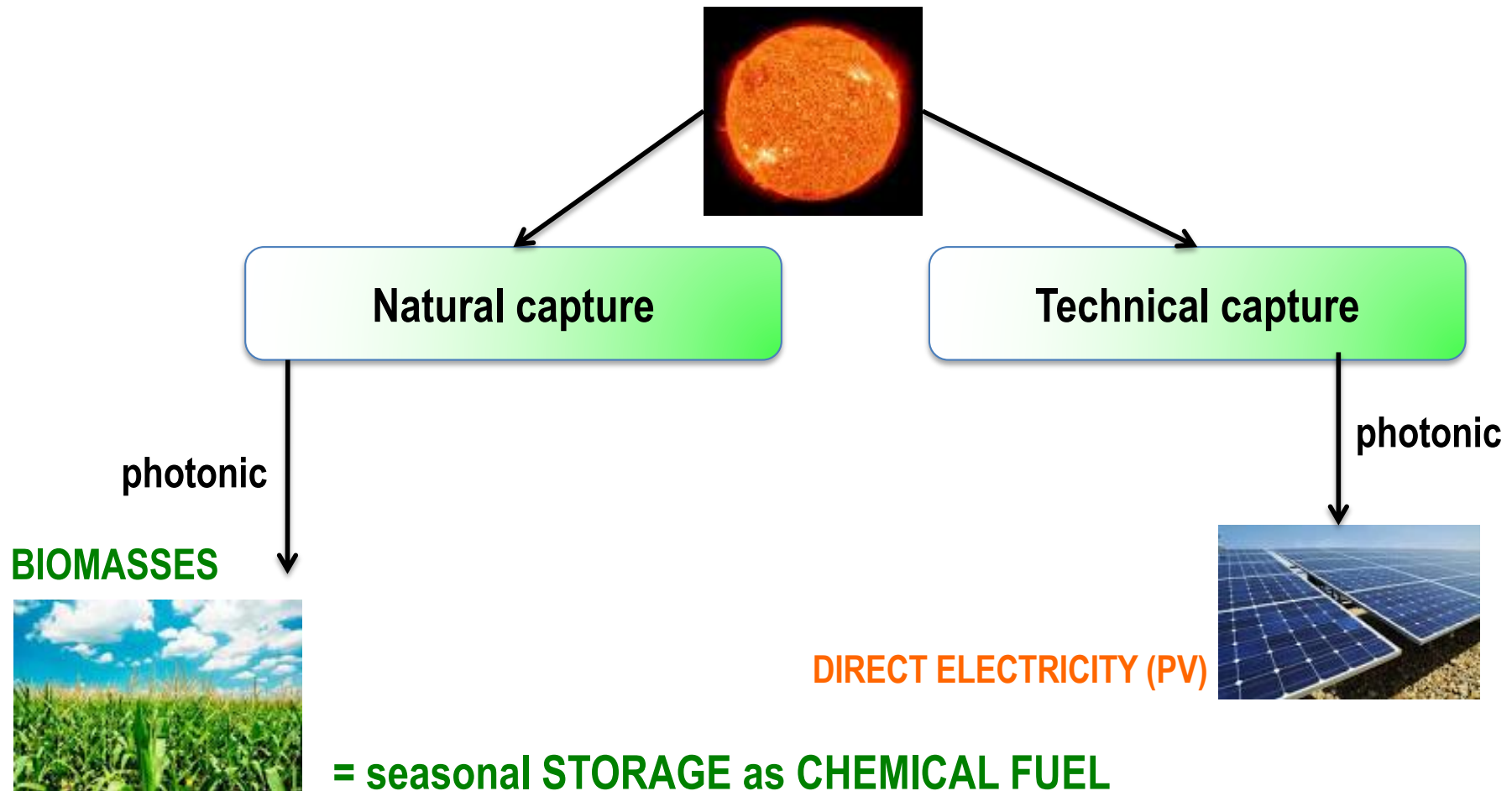


Biomass: resources & conversion technologies



Biomass course part: overview

- Definitions
 - Photosynthesis, compositions, structure
- Potential : theoretical vs. real
- Conversion
 - 1. Solids** (wood; energy crops)
 - combustion
 - pyrolysis
 - gasification
 - solids-derived liquid/gaseous fuels (=‘secondary’ biofuels)
 - 2. Liquids** (bioethanol; biodiesel)
 - fermentation
 - extraction
 - application as ‘primary’ biofuels (engines)
 - 3. Gas** (biogases)
 - anaerobic digestion

Learning objectives

- Know and **distinguish the types** of biomasses (as well as the appropriate **conversion route** per biomass type)
- Know theoretical biomass potential (photosynthesis efficiency) and the estimates of real biomass **potential**
- Quantify the 'energy vs. **food**' competition for biomass resource
- Explain advantages (& drawbacks) of biomass as energy carrier, in particular for **residual** biomass
- Know approximately the chemical **structure** of biomass ('ligno-cellulose')
- Estimate the **LHV** of a biomass from its composition

Theoretical photonic (solar) capture potential

PHOTOSYNTHESIS



Extraterrestrial radiation arriving at the Earth's outer atmosphere:

5.5 E+24 J/yr (=1368 W/m², solar constant)

Solar radiation on Earth's surface (where vegetation can capture it)

averages out on a yearly basis to **5.1 E+23 J/yr**

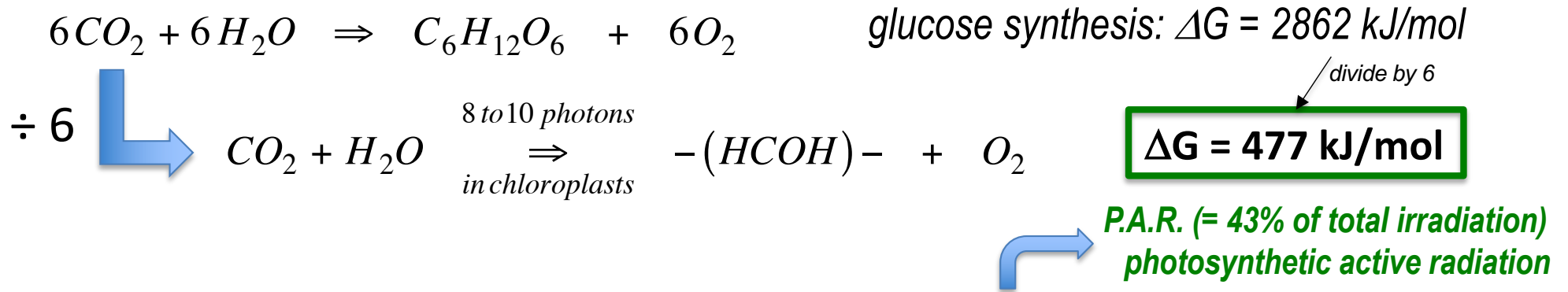
(which is ~160 W/m² or 5 GJ/m².yr ≈ 1400 kWh/m².yr)

Compared to the world annual primary energy:

5.4 E+20 J/yr (= 540 EJ = 13 Gtoe)

→ theoretical potential ≈ **1000 times** the human primary energy need

Theoretical photosynthesis efficiency



Considering red (660 nm), blue (450 nm) or white light (400-700 nm: the visible spectrum), and Planck's formula,

$$E_{\text{photon}} = h \cdot \nu = h \cdot \frac{c}{\lambda}$$

$h = 6.62 \cdot 10^{-34} \text{ Js}$: Planck's constant
 $c = 3 \cdot 10^8 \text{ m/s}$: speed of light

photon energy amounts to $\frac{2 \cdot 10^{-16}}{\lambda \text{ (in nm)}} \text{ J/photon}$

Hence for 1 mol of photons ($\times 6.02 \cdot 10^{23}$) $\rightarrow \frac{120}{\lambda \text{ (in nm)}} \text{ MJ/mol}$

- \rightarrow red (660 nm): 181.2 kJ/mol
- \rightarrow blue (450 nm): 265.6 kJ/mol
- \rightarrow white (577 nm): 208.1 kJ/mol

Theoretical P.S. efficiency (white light) is then, considering 8 photons for -HCOH- synthesis:

$$\frac{\text{glucose output}}{\text{photonic input}} = \frac{477 \text{ kJ/mol}}{8 \text{ photons} \cdot 208.1 \text{ kJ} / 1 \text{ mole of white light photons}} = 28.6\%$$

Real biomass photosynthetic 'efficiency'

Process / Maximal solar input on ground level: 1400 kWh/m ² .yr or 160 W/m ² =	100%
↓	
Solar radiation energy → photosynthetic active part, PAR (400-700 nm)	43%
Maximum capture by leafs (canopy) = 80% (effective square meters available)	34.4%
Maximum photonic energy capture efficiency into glucose = 28.6% (p. 5)	9.8%
1/3 on average of the glucose energy is used for the plant metabolism (respiration)	6.6%
Max. practical efficiency of 'C-4' 'energy' plants (corn, sorghum, sugar cane), on daily basis (24h)	5%
Max. practical efficiency of 'C-3' common plants (=95% of biomass, e.g. wheats, rice, trees,...), on daily basis (24h)	3%
→ from the available 5.1 E+23 J/yr radiation (1400 kWh/m ² .yr), thus 3% is theoretically captured by common biomass (42 kWh/m ² .yr = 150 MJ/m ² .yr)	1.5 E+22 J/yr (4.8 W/m ²)
Climate factors, shading, and biomass density per m ² drop this capture efficiency by another factor ~5 (→ 1 W/m ² = 8.4 kWh/m ² .yr = 30 MJ/m ² .yr ≈ 2 kg wood/m ² .yr)	0.6% 3 E+21 J/yr

1 W/m² is a poor storage density! (20 tonnes (dry) / hectare.yr)

Even for a '2 kW-society', every citizen would need his personal 2000 m² 'storage' surface

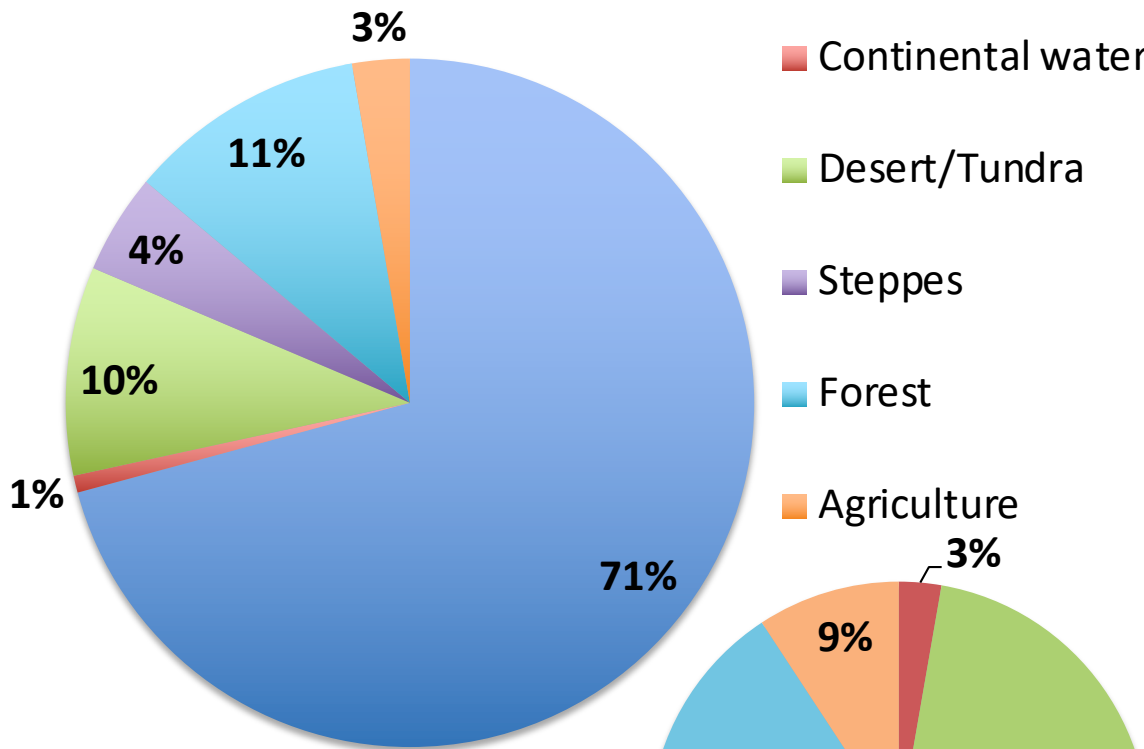
p.8

Biomass production of the biosphere

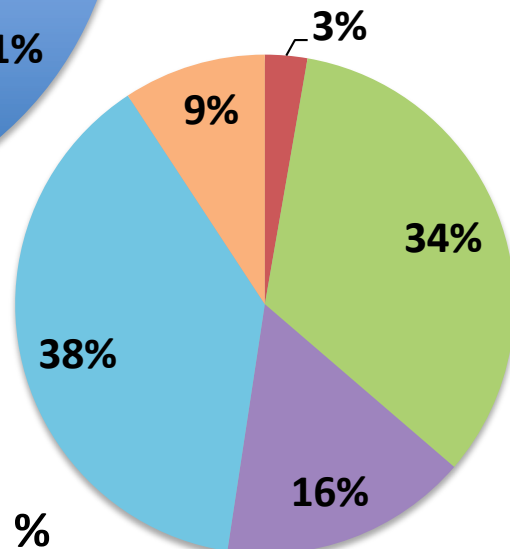
$3 \cdot 10^{21} \text{ J/yr} = 3000 \text{ EJ} \approx 70 \text{ Gtoe} \approx 95 \text{ TW} \approx 5 \times \text{the world annual primary energy}$

≈ 10 toe biomass energy per person on the planet!

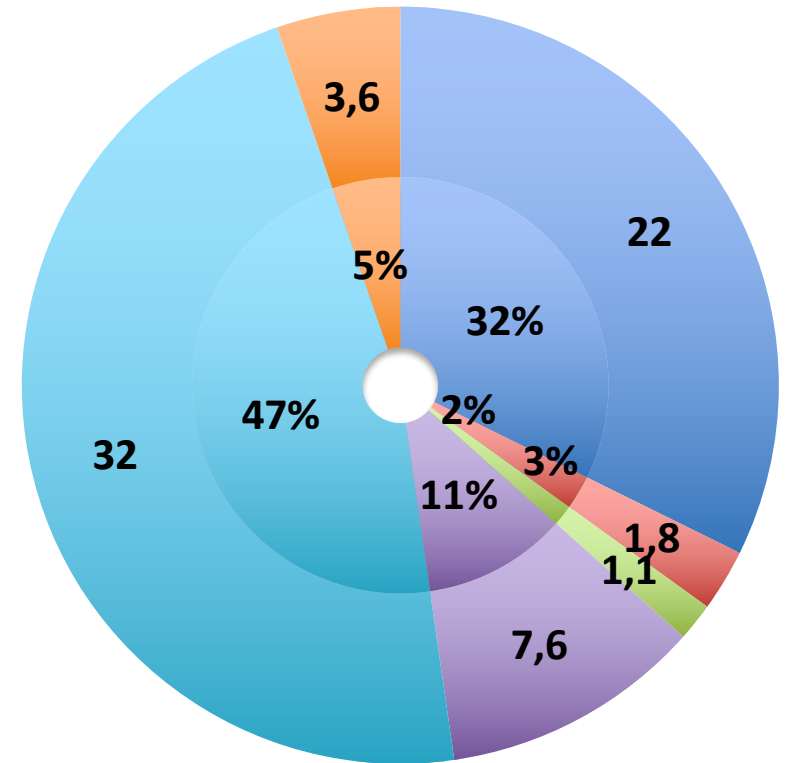
GLOBE AREA (%)



land area %



in Gtoe



⇒ 'wood': 2.5 x the world annual energy
⇒ 'agro': 0.3 x the world annual energy

Sustainable biomass potential

- Primary production of biomass in the biosphere ($3 \cdot 10^{21}$ J)
 $\approx 200 \cdot 10^9$ tonnes (dry) /yr (assuming 15 MJ per kg dry biomass)
- Theoretically exploitable : 57% (without oceans, desert,..)

- **Technically sustainable*** : ca. **9%**

= **agriculture (5%)** + ~10% of forestry (47%), cf. previous slide
($\approx 4\%$ of the Earth's total surface, or $\approx 13\%$ of the emerged lands)

= $18 \cdot 10^9$ tonnes (dry) / yr

= $270 \cdot 10^{18}$ J = 270 EJ (6.4 Gtoe)

= **50% of world annual primary energy (540 EJ)**

(*>half of which (150 EJ, 3.6 Gtoe) is used/meant for food, mainly*)

Influence factors: :

 An important part can be recovered as energy from the residues

nutrition, moisture, CO₂ concentration, light, temperature, leaf anatomy,..

→ in practical terms, the sustainable biomass energy potential could amount up to $\approx \frac{1}{3}$ (180 EJ) of the present human energy needs.

The main source is wood (>120 EJ) and the remainder from other biomass sources, an interesting source being residual biomass (i.e. 'waste streams' – cf. further below)

Examples of real biomass yield

(in case of 0.6% efficiency = 30 MJ/m².yr = **300 GJ/ha.yr**; 1 ha = 1 hectare = 10'000 m²)

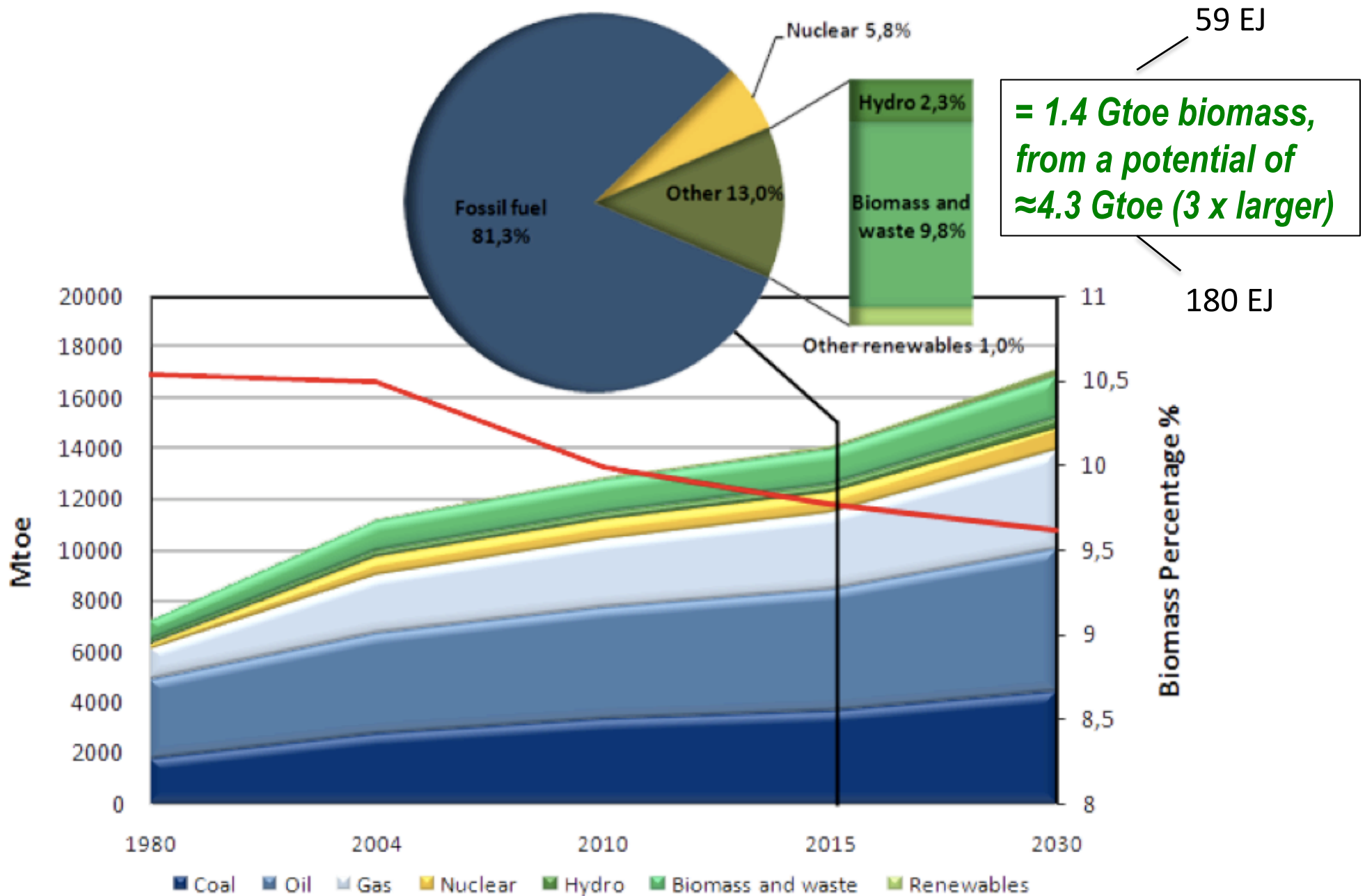
Plant	Energy output (GJ/ha/year)
1. Switchgrass	185...215
2. Miscanthus	Up to 785 (calc. with LHV=17,8 MJ/kg)
3. Sugar beet	62...96
4. Rape seed	14...78
5. Sweet sorghum	54...58
6. Wheat	23...47
8. Wood (forest)	100 (calc. on 50% dm with 10 t/ha yield)

***real yield ≈
typically even
only 10-30%
of the raw
photosynthetic
yield !***

Sources:

- 1) 1: I.C. Madakadze et al.: Light interception, use-efficiency and energy yield of switchgrass (*panicum virgatum* L.) grown in a short season area. Biomass and Bioenergy, vol 15, No. 6, pp. 475-482, 1998
- 2) 2: I. Lewandowski et al.: Miscanthus: European experience with a novel energy crop, Biomass and Bioenergy 19 (2000) 209-227
- 3) 3-6: P. Venuri, G. Venuri: Analysis of energy comparison for crops in European agricultural systems. Biomass and Bioenergy 25 (2003) 235-255
- 4) Phyllis database available at <http://www.ecn.nl/phyllis/>

Biomass exploitation reality



Is there competition with food? (exercise)

- An adult human being is a **120 W** machine. Assume we get our energy from 80% vegetables (= '*direct*' biomass) and 20% from meat (= '*indirect*' biomass). (Assume efficiency from primary biomass-to-meat = 10%)
- How much MJ/day, and kWh/yr, do you need in food from primary biomass?
- How much primary biomass does the world consume in this way? (7.5 billion people)
- Discuss the result in view of the biomass potential for energy, and current agricultural production.

Motivation for biomass use as **energy** resource

- the primary yearly biomass production (3000EJ) is >5-fold the total world human primary energy consumption (550 EJ)
- agricultural land is <10% of the total land area;
agricultural production = 5% of the biosphere energy production (152 EJ); this is enough to feed the planet, leaving residual energy
- optimised **cultivation** can raise the **effective photosynthetic efficiency** above the average value of 0.6% (=30 MJ/m².yr); the theoretical limit is 3% to 5% storage efficiency for C3 and C4 plants (i.e. a maximal potential up to 100-250 MJ/m².yr)
- **marginal land** areas can be used for 'energy cultures'
- technologies for production and conversion are relatively well **established** or developed
- **CO₂ neutral**, and less overall polluting emissions (vs. fossil)

Biomass use for energy

Advantages

- renewable
- $\approx 100\%$ use of collected matter
- rel. conventional technologies
- environmentally benign
- employment, labour intensive
- fuel import savings
- energy supply security

Drawbacks

- dispersed resource
- seasonal production
- low energy density
- requires transport and storage means
- some of the transformations involved are cumbersome (mechanical and chemical treatments,...)

Biomass classification by **water** content

- 'dry' < 15 wt% humidity
- 'humid' 15-30 wt% H₂O
- '**slurry**' 30-90 wt% H₂O (without 'structure')
 - e.g. animal manure
 - e.g. 'molasse' (=the sirupy byproduct from sugar plants)
- '**liquid**' > 90 wt% H₂O
 - waste waters
 - sewage
 - industrial effluents with 'high' organic charge (e.g. food industry)

Biomass classification by **human** activity

- **natural** biomass (=protected areas, no human interference)
- **residual** biomass (=organic waste streams from human activity)
 - **passive use** of biomass ('recovered' as energy resource)
 - agricultural residues
 - forestry maintenance
 - animal breeding / farming
 - industry (industrial solid and liquid wastes, e.g. food industry)
 - urban centers (municipal solid waste; sewage)
- **cultivated** biomass → **active use** of the land for energy
 - agricultural excess (e.g. non-edible parts of the harvest)
 - 'energy crops' (non-food)

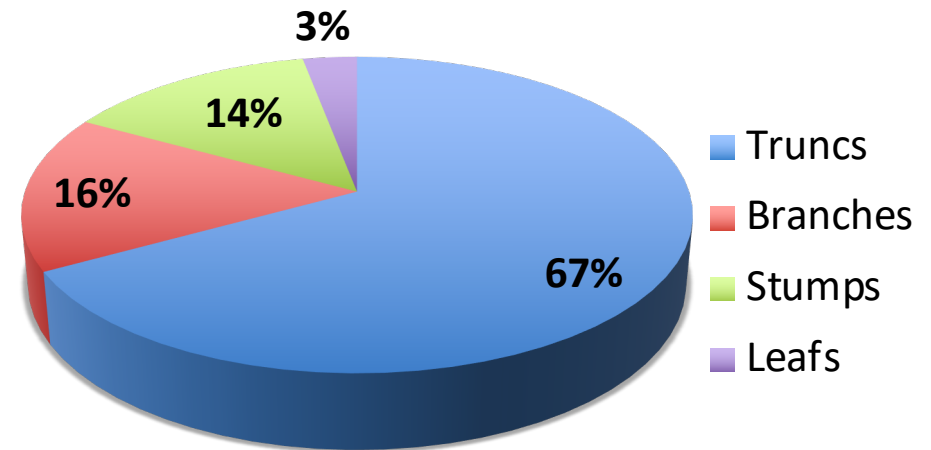
Residual biomasses (= 'waste streams' from human activity)

- **Agriculture residues**

- cereals
- fruit trees, vineyards, olive trees (lignic)
- industrial crops (oily plants)

- **Forestry**

- trimming residues
- wood industry
 - sawdust, bark, shavings
- forestry maintenance (1 kg/m²)



- **Animal breeding**

- slaughterhouses
- manure

- **Industry** (solids, liquids: effluents with organic charge)

- **Public waste** (municipal solid waste MSW; sewage)

(estimate for liq. wastes = 150-300 L/day/person containing 0.4 kg organic dry solids)

Estimate of *residual* biomass, primary and final energy = energy recovered from waste streams (exercise)

Assumptions / Conversion factors:

1. **agriculture residues**: from total production (152 EJ), discount human food requirement (cf. exercise p.13). Assume that from the remainder, $\approx\frac{1}{2}$ is used to feed animals, $\approx\frac{1}{4}$ is used for composting, and the rest (assume 10%) is recoverable as energy
2. **forestry**: assume 1 kg/m² per year of dry wood (LHV:17 MJ/kg); assume 1% of the world's forests area is trimmed (from where this 'waste wood' is recovered)
3. **animal manure**: assume a production of 1 m³ of biogas per day (with 50% CH₄ content) per large farm animal and there are half as many large farm animal- 'equivalents' as people.
4. **solid organic wastes** from our activities (kitchen waste, park&garden waste, food industry): assume 1 kg dry organic matter waste per week per person, converted to 500 L biogas per kg dry waste, with a CH₄ content of 60%
5. **human liquid organic waste** (sewage): assume a production of 30 L biogas per day per person, with a CH₄ content of 65%
6. Finally, you need to assume realistic conversion efficiencies from primary to final energy (whether heat or power) for the different sources!

***Residual* biomass: advantages**

- **low cost** production (can even be zero or *negative* cost ‘fuel’)
- closed cycle: minerals (inorganic part) are reused for **fertilising**
- **local** exploitation (= low transport cost)
- reduced contamination or load on waste management
- ‘free’ energy recovery, which amounts to at least several % (and up to 10%) of total energy needs !
 - e.g. for Switzerland: 56 PJ incinerated solid wastes (of which $\approx 50\%$ is considered renewable), 40 PJ wood use and 5 PJ of biogas (both of which are much *underexploited*) = total of 100 PJ = 8% of Swiss primary energy need

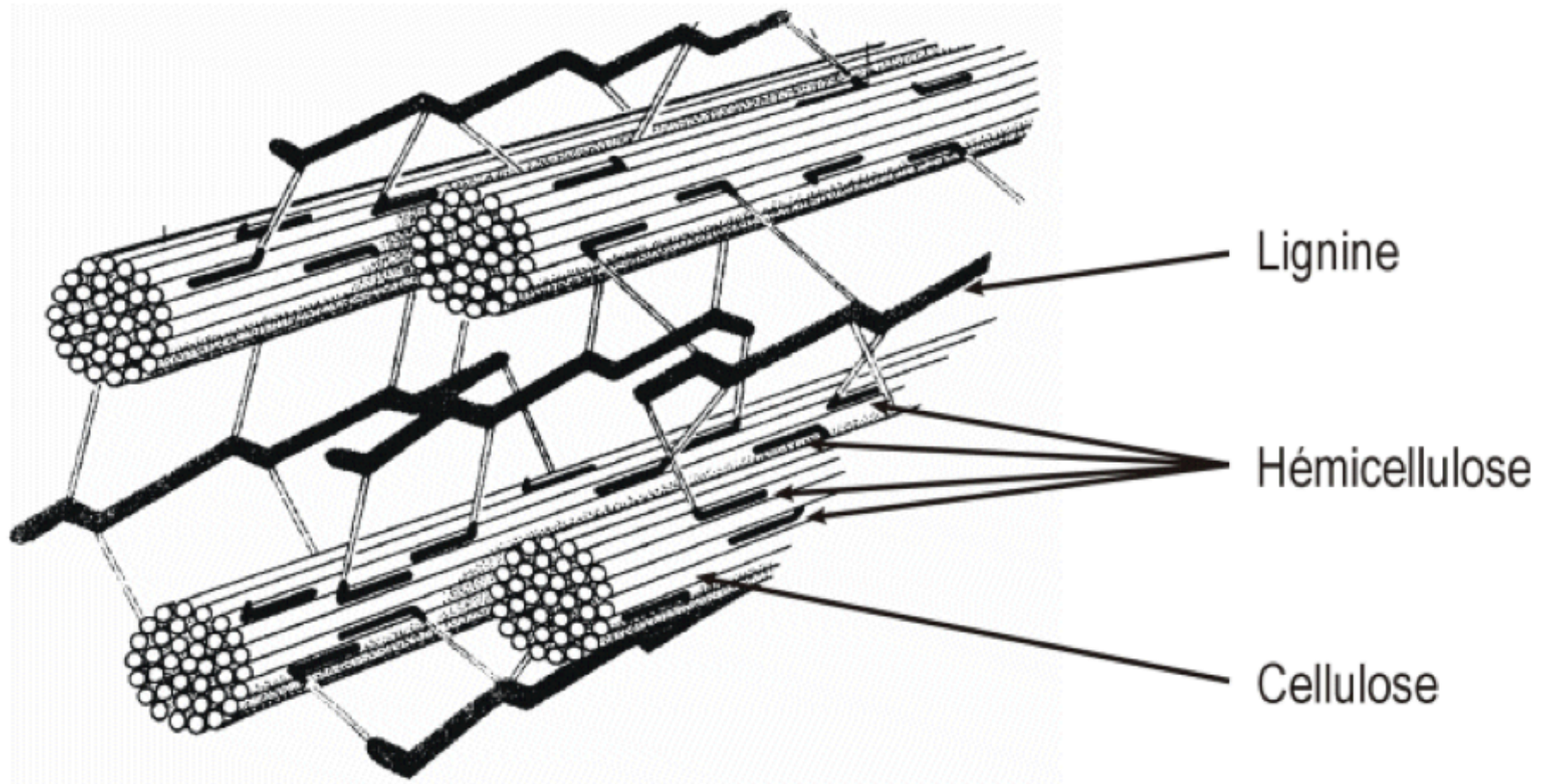
'Residual' biomass energy: Swiss case

- 56 PJ incinerated solid wastes (MSW/ISW, waste wood; in part NG-assisted)
(*Remark: only ≈50% of this is in fact renewable (rest = fossil origin, mainly plastics)*)
- 40 PJ indigenous wood use (=> potential could be doubled)
- 5 PJ of biogas (largely *underexploited*)
=> could be increased >5-fold (≈30 PJ)

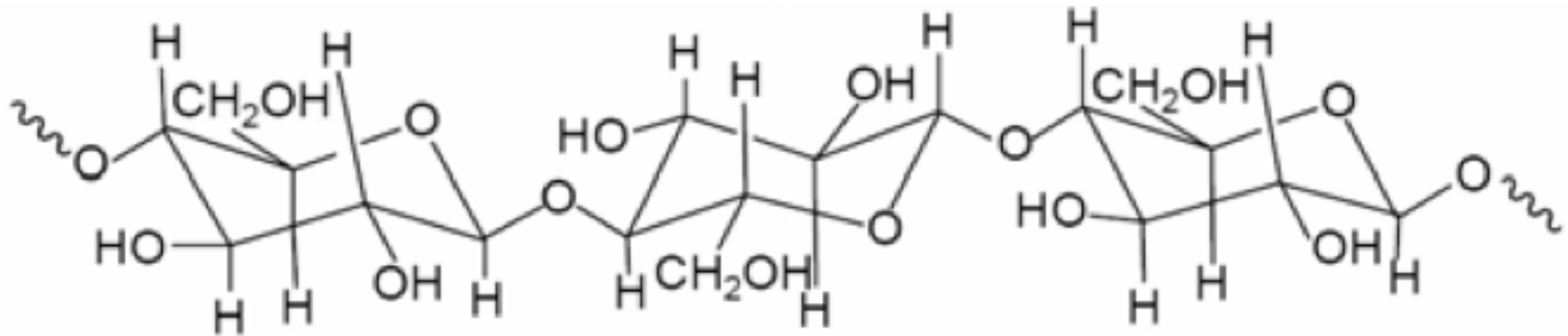
= present total of 100 PJ = 8% of Swiss primary energy

- ca. 6.5% of final energy; **5% of Swiss electricity (as renewable: 3%)**
- electricity 10.2 PJ (20% efficiency) from incinerated solid wastes (2.85 TWh); in addition 30% heat is produced and distributed as district heat
- electricity 1.2 PJ (27% efficiency) from biogases (0.3 TWh) => **0.5% of total electricity**
- electricity production from wood is negligible

Ligno-cellulosic biomass structure



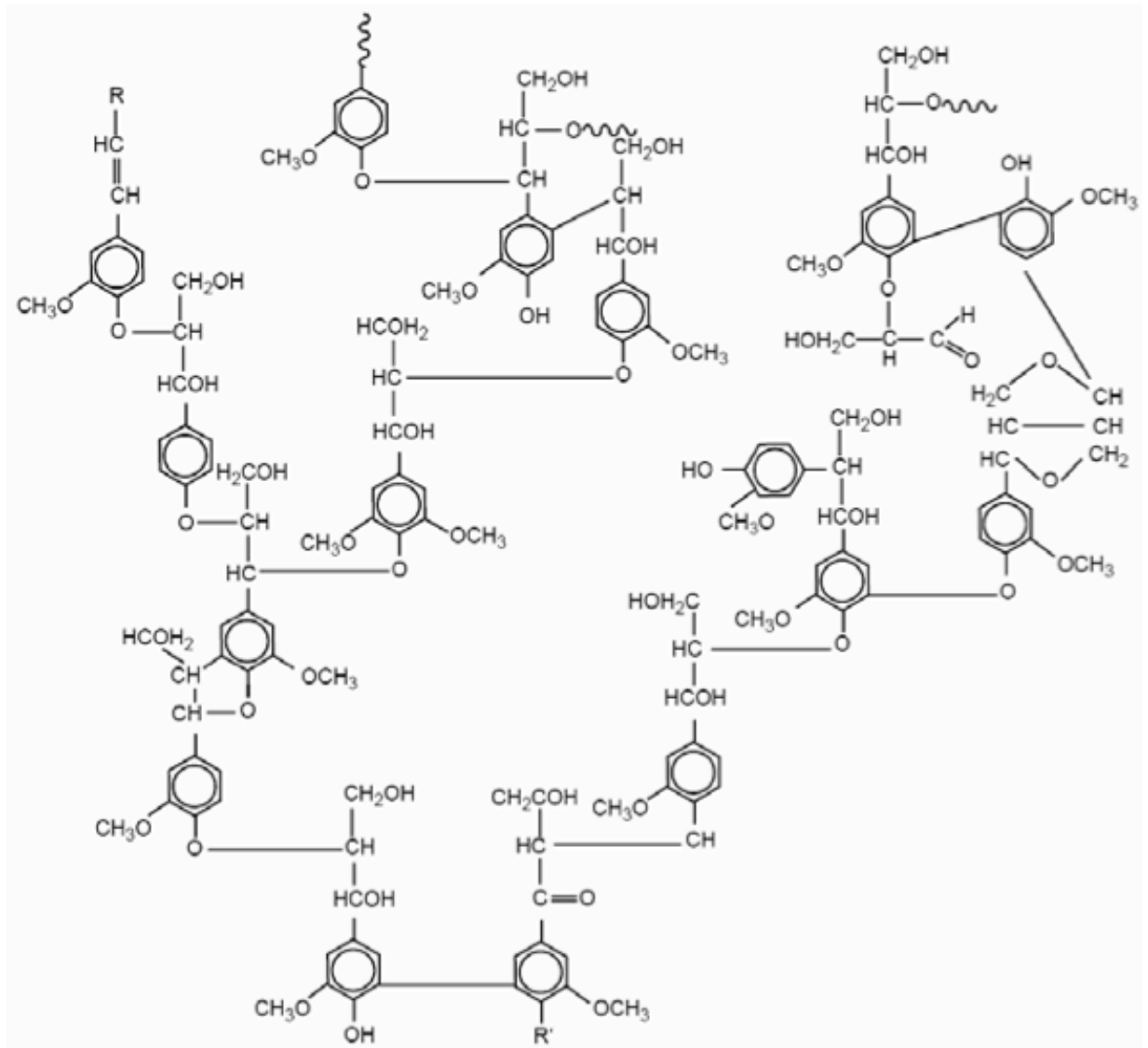
Cellulose



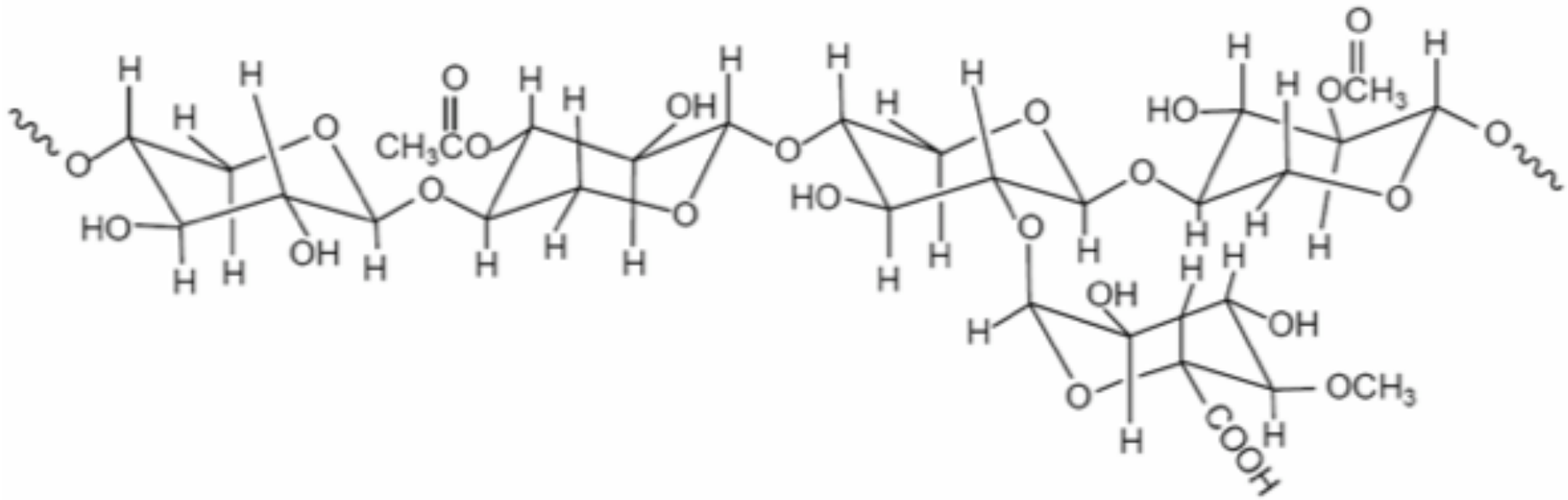
- 40-80 wt% in plants, **17.5 MJ/kg** (C:H:O \approx 30:45:25 at%)
- '**soft**' part in plants
- linear polymer of up to 10'000 glucose (C₆) molecules:
 $(C_6H_{10}O_5)_n$

Lignine

- complex **aromatic** polymer
 - ca. $(C_{10}H_{12}O_4)_n$
 - 25-35 wt% in wood
 - 10-25% in plants
 - responsible for **slow growth and rigidity**
 - **26.6 MJ/kg**
- (C:H:O \approx 40:45:15 at%)



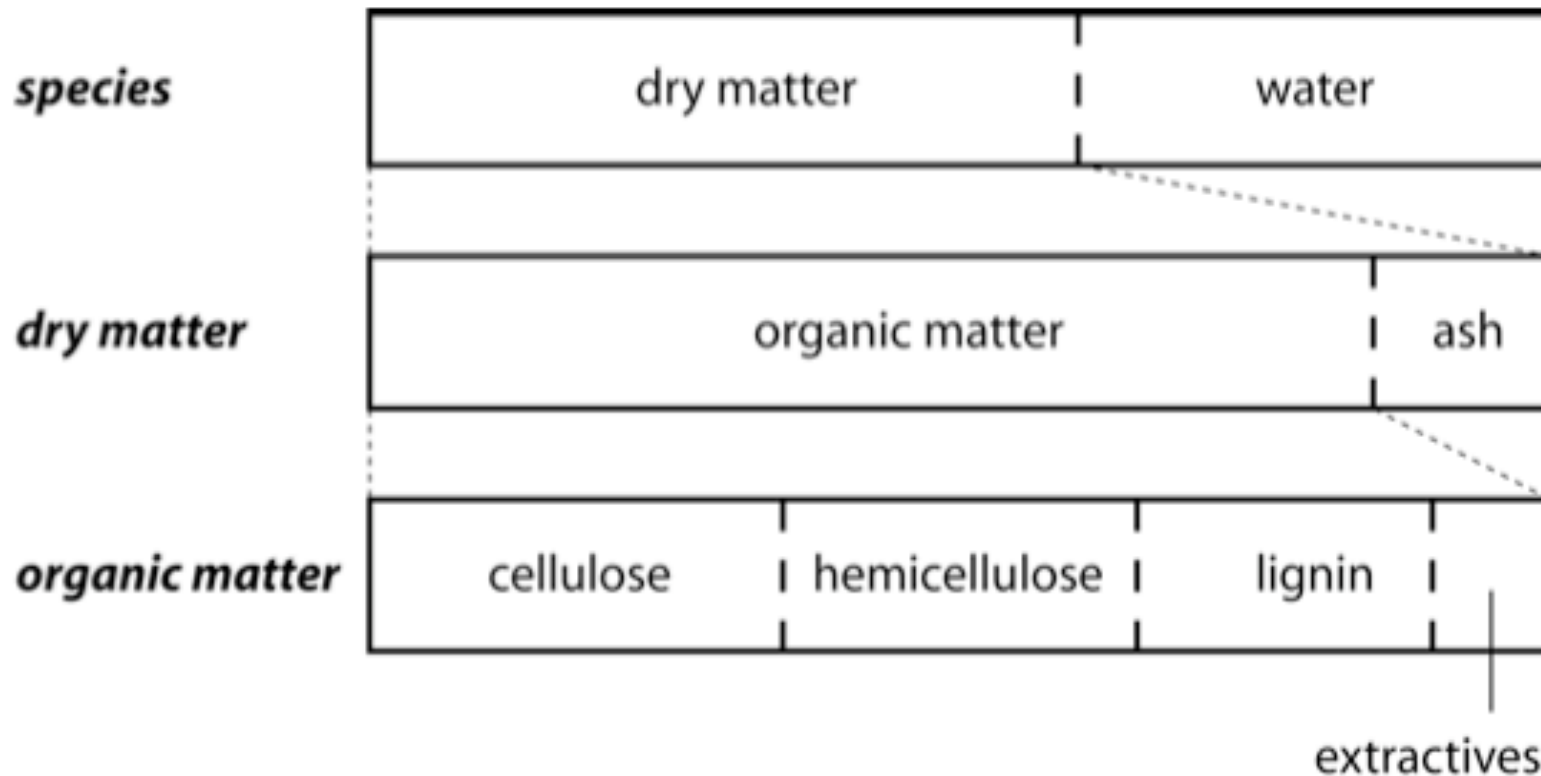
Hemi-cellulose (xylose)



- 15-30wt% of plants, C₅H₈O₄, **17.5 MJ/kg** (C:H:O ≈ 30:45:25 at%)
- **'connects'** lignine to cellulose
- 'shorter' polymer of 50-200 sugar molecules (C5 structures)
- 5 sugars: xylose, arabinose, galactose, glucose, mannose

Structural composition

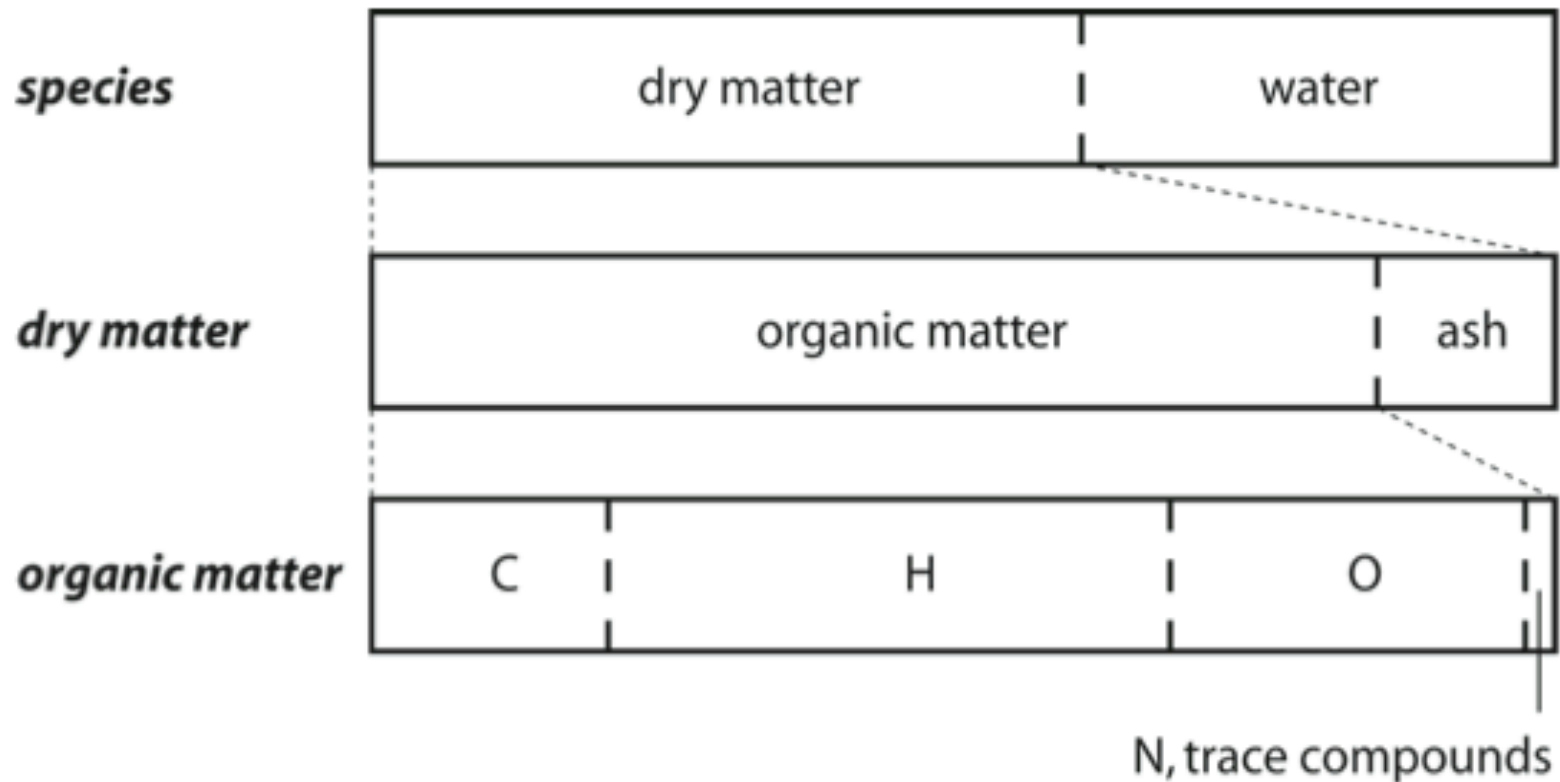
macromolecular description:



cellulose	31-47%wt	lignin	8-31%wt
hemicellulose	24-35%wt	extractives	1-31%wt

Chemical composition

atomic description:



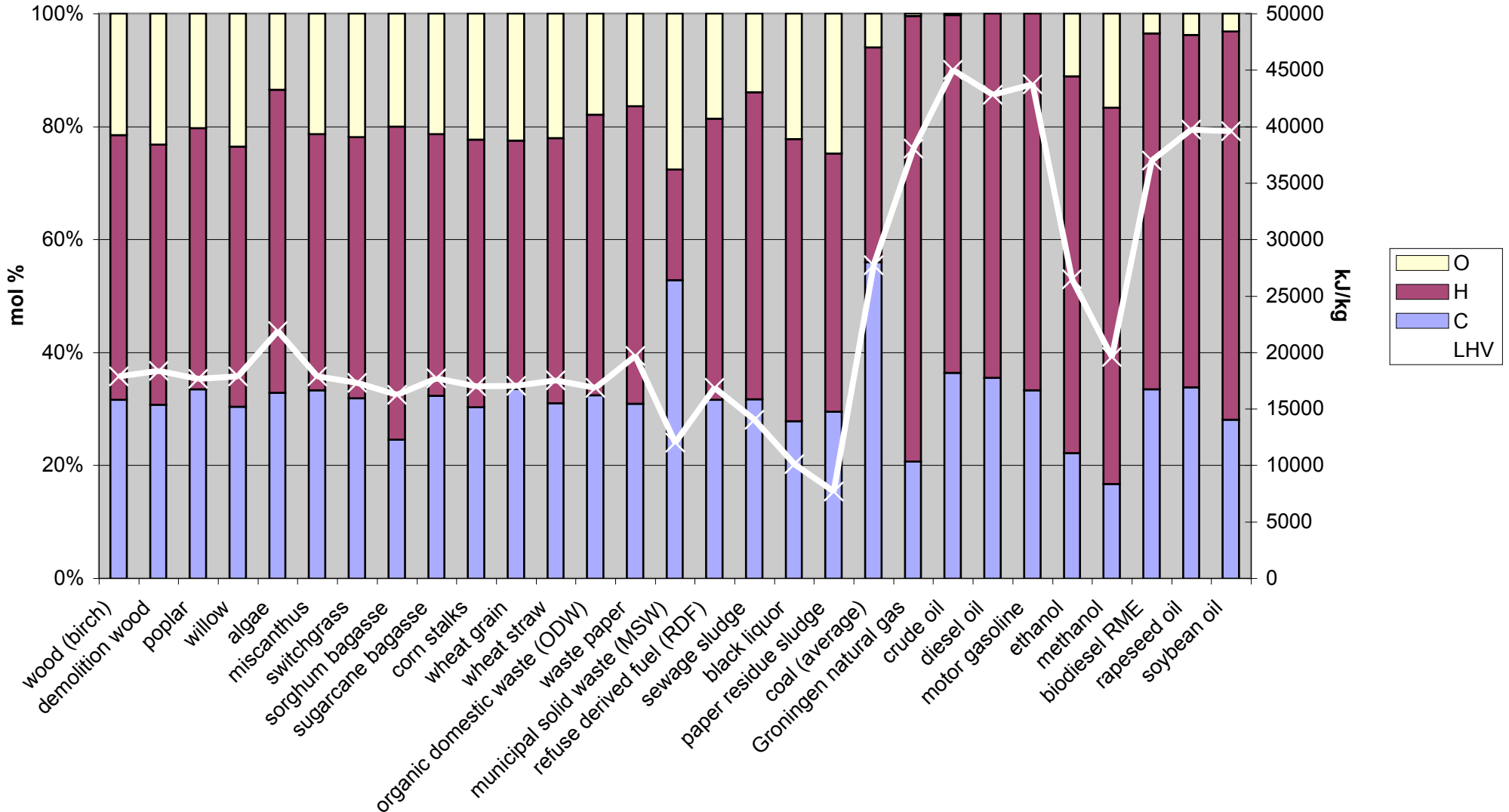
C	~ 30%mol	O	~ 20%mol
H	~ 50%mol	N	~ 1%mol

'Dry' wood (with 11% humidity)

Element	Weight %
C	47
H	6
O	35
N	0.1
S	0.0
Ash	1
Water	11

Composition and energy content of fuels

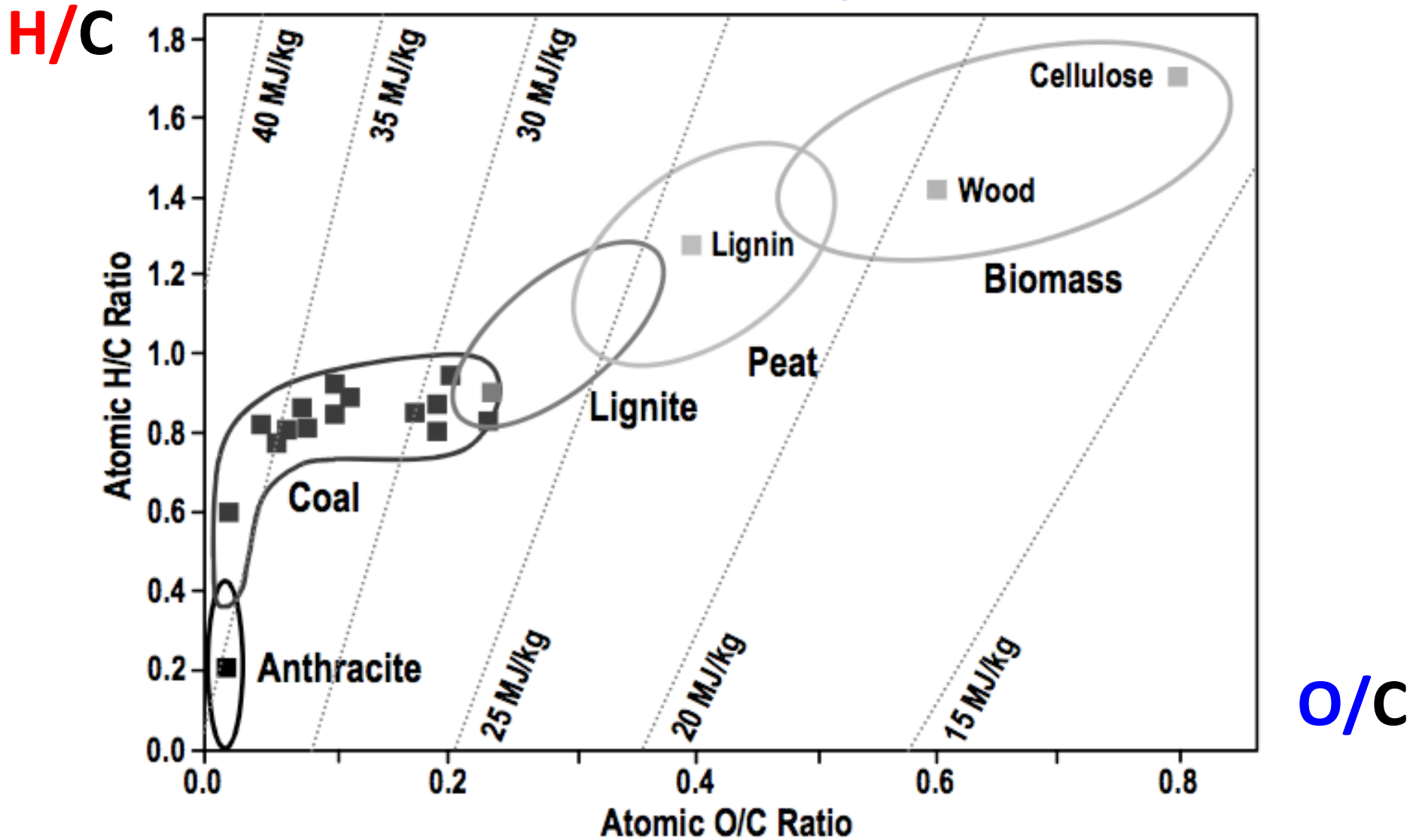
Elementar composition of biomass, fossil fuels and biofuels and lower heating value



The **OXYGEN** content in biomass fuels (=OXY-HYDROCARBONS) reduces the LHV to **<20 MJ/kg** compared to **>40 MJ/kg** for (fossil) HYDROCARBONS

Heating value and C/H/O composition

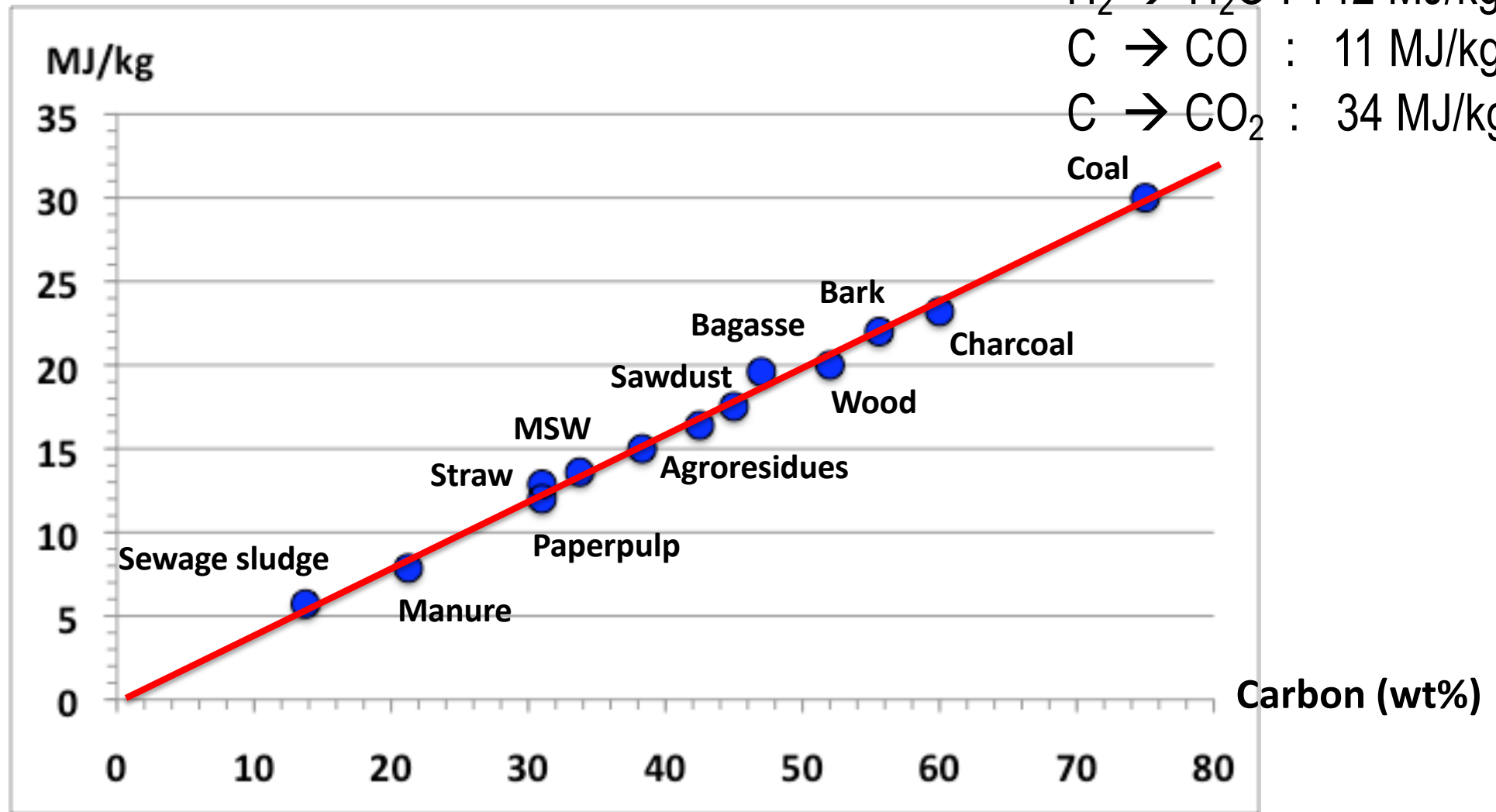
Van Krevelen diagram



Heating value vs. carbon content

Reminder:

$H_2 \rightarrow H_2O$: 142 MJ/kg
 $C \rightarrow CO$: 11 MJ/kg
 $C \rightarrow CO_2$: 34 MJ/kg



$LHV = 43.6 * C - 0.31 \quad MJ/kg$ where $C = wt\% \text{ carbon } (<80)$

Compositions of biomasses

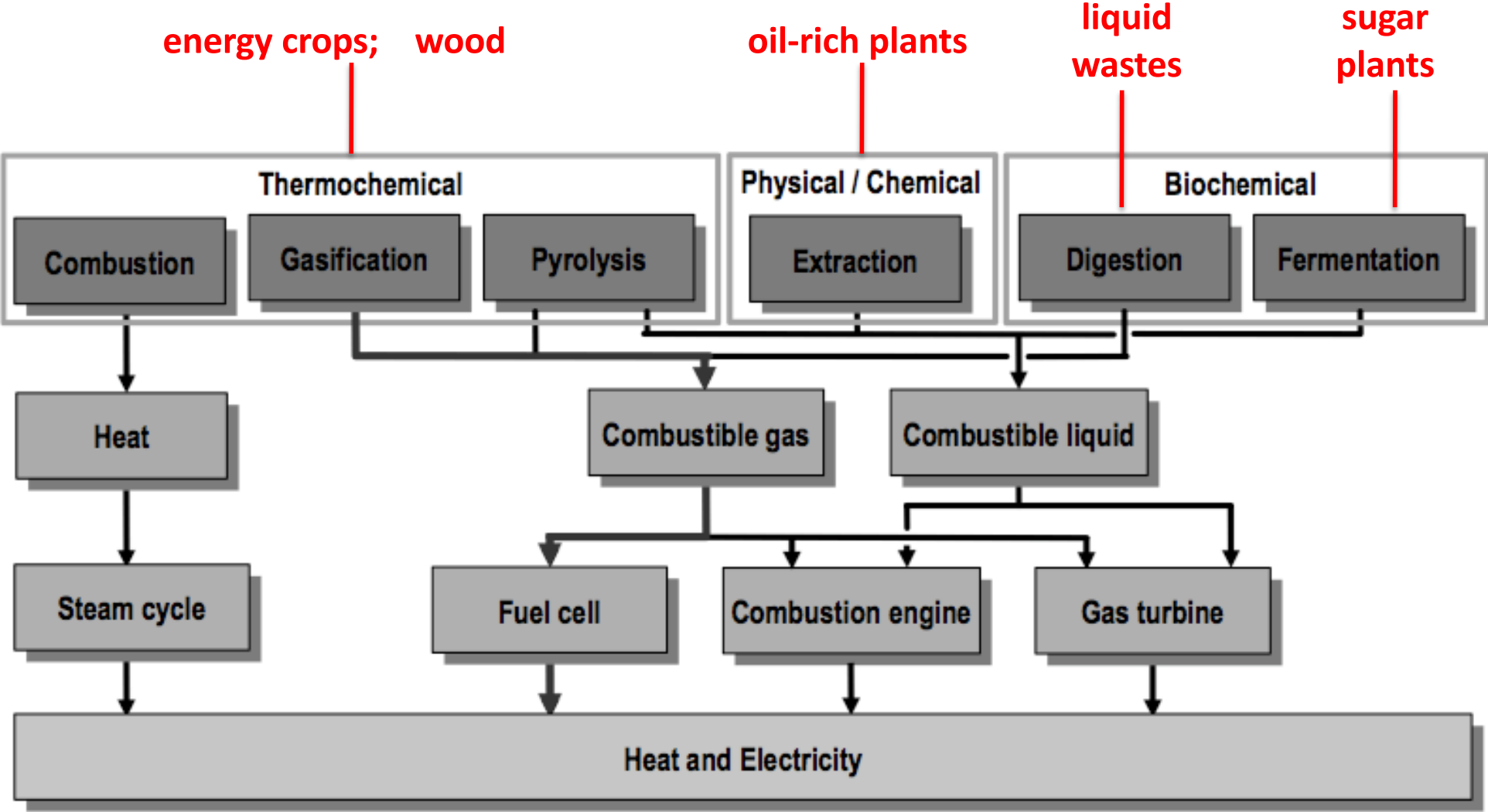
Source	C	H	O	N	S	Inorg.	LHV MJ/kg
carbon	100						29.3
coal	70-80	5	5-20	1-1.5	1-3	4-15	30-34
wood	52	6	40	0.1	0	1	21
bagasse	47	6	35	0	0	11	21
untreated sewage	45.5	7	26	2.4	0.5	19	16.4
cattle manure	42.7	5.5	31	2.4	0.3	18	17
rice residue	39	5.4	38	0.5	0	18	15
MSW	34	4.6	22	0.7	0.4	38	13
paperpulp	31	7	51	0.5	0.2	10	12
sewage sludge	14	2	11	1	0.7	71	5

$$LHV = 43.6 * C - 0.31 \text{ MJ/kg} \quad \text{where } C = \text{wt\% carbon } (<80)$$

The carbon content alone is a reasonable measure for the heating value.

As if the LHV (expressed per kg fuel) gain due to H were 'lost' due to the presence of O mass in the fuel.

Biomass conversion schemes overview



F. Nagel (PSI)

BIOMASS CONVERSION ROADMAP

